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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/798,628	03/12/2004	Harry E. Flynn	2019	6686	
7:	590 06/13/2006		EXAMINER		
William B. Miller			FIORITO, JAMES		
Kerr-McGee Corporation 123 Robert S. Kerr Avenue			ART UNIT	PAPER NUMBER	
Oklahoma City, OK 73102					
				DATE MAILED: 06/13/2006	

Please find below and/or attached an Office communication concerning this application or proceeding.

		Application No.	Applicant(s)				
Office Action Summary		10/798,628	FLYNN ET AL.				
		Examiner	Art Unit				
		James A. Fiorito	1754				
T Period for R	he MAILING DATE of this communication app eply	ears on the cover sheet	with the correspondence address				
WHICHE - Extension after SIX - If NO peri - Failure to Any reply	TENED STATUTORY PERIOD FOR REPLY EVER IS LONGER, FROM THE MAILING DAIS of time may be available under the provisions of 37 CFR 1.13 (6) MONTHS from the mailing date of this communication. od for reply is specified above, the maximum statutory period we reply within the set or extended period for reply will, by statute, received by the Office later than three months after the mailing stent term adjustment. See 37 CFR 1.704(b).	TE OF THIS COMMUN (6(a). In no event, however, may fill apply and will expire SIX (6) Mic cause the application to become	IICATION. a reply be timely filed ONTHS from the mailing date of this communic ABANDONED (35 U.S.C. § 133).				
Status							
1)⊠ Re	esponsive to communication(s) filed on 12 M	arch 2004.					
2a)∐ Th	This action is FINAL . 2b)⊠ This action is non-final.						
clo	sed in accordance with the practice under E	x parte Quayle, 1935 C	.D. 11, 453 O.G. 213.				
Disposition	of Claims						
4)⊠ Cla	aim(s) <u>1-19</u> is/are pending in the application.						
4a)	4a) Of the above claim(s) is/are withdrawn from consideration.						
5)∐ Cla	Claim(s) is/are allowed.						
•	Claim(s) <u>1-19</u> is/are rejected.						
-	aim(s) is/are objected to.						
8)[_] Cla	aim(s) are subject to restriction and/or	election requirement.					
Application	Papers						
9) <u></u> The	e specification is objected to by the Examine	г.					
10)⊠ The	e drawing(s) filed on <u>12 March 2004</u> is/are: a	a)⊠ accepted or b)□ c	bjected to by the Examiner.				
•	plicant may not request that any objection to the	• • • • • • • • • • • • • • • • • • • •					
	placement drawing sheet(s) including the correct						
11)∐ The	e oath or declaration is objected to by the Ex	aminer. Note the attach	ed Office Action or form PTO-15	02.			
Priority und	ler 35 U.S.C. § 119						
12) <u></u> Acl a)∏ <i>i</i>	knowledgment is made of a claim for foreign All b) ☐ Some * c) ☐ None of:	priority under 35 U.S.C	. § 119(a)-(d) or (f).	•			
· · · · ·	Certified copies of the priority documents	s have been received.					
2.	2. Certified copies of the priority documents have been received in Application No						
3.	Copies of the certified copies of the prior	rity documents have bee	en received in this National Stage	е			
	application from the International Bureau	ı (PCT Rule 17.2(a)).					
* See	the attached detailed Office action for a list	of the certified copies n	ot received.				
Attachment(s)							
	References Cited (PTO-892)		w Summary (PTO-413) lo(s)/Mail Date				
3) X Informat	f Draftsperson's Patent Drawing Review (PTO-948) ion Disclosure Statement(s) (PTO-1449 or PTO/SB/08) b(s)/Mail Date <u>3/12/2004</u> .		of Informal Patent Application (PTO-152)				

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DETAILED ACTION

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1-2, 4-7, 12 and 18 are rejected under 35 U.S.C. 102(b) as being anticipated by Weaver '049.

Weaver discloses a process for producing particulate solid titanium dioxide comprising: (a) reacting gaseous titanium tetrachloride and oxygen to produce solid particulate titanium dioxide and gaseous reaction products in an oxidation reactor; (b) quenching the particulate titanium dioxide and gaseous reaction products with a recycled stream of gaseous reaction products which has been previously cooled by injecting a portion of the cooled recycled stream of gaseous reaction products into a zone in the reactor where the reaction is complete and titanium dioxide primary particles are no longer growing in size, said recycled gaseous reaction products being injected at a pressure of less than 75 psig above the reactor pressure, and at a temperature significantly less than the reactor temperature at the zone of injection; (c) cooling the quenched particulate titanium dioxide and gaseous reaction products in a tubular heat exchanger; (d) separating the cooled particulate titanium dioxide from the cooled gaseous reaction products, from which the titanium dioxide has been removed as in step (d), to the

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reactor to provide the quench as called for in step (b) (Column 1, line 8-5; Column 2, line 54-69; Column 3, line 31-74).

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-2, and 4-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hilgers '168.

Hilgers discloses a process for producing particulate solid titanium dioxide comprising: (a) reacting gaseous titanium tetrachloride and oxygen to produce solid particulate titanium dioxide and gaseous reaction products in an oxidation reactor; (b) quenching the particulate titanium dioxide and gaseous reaction products with a recycled stream of gaseous reaction products which has been previously cooled by injecting a portion of the cooled recycled stream of gaseous reaction products into a zone in the reactor where the reaction is complete and titanium dioxide primary particles are no longer growing in size, said recycled gaseous reaction products being injected at room temperature at the zone of injection; (c) cooling the and gaseous reaction products; (d) separating the particulate titanium dioxide from the gaseous reaction products in a pipe; and (e) recycling a portion of the cooled gaseous reaction products,

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from which the titanium dioxide has been removed as in step (d), to the reactor to provide the quench as called for in step (b) (Column 9, Example 3).

Hilgers does not expressly state that the quenched particulate titanium dioxide is cooled. However, it would have been obvious to cool the particulate in order to more safely ship and store the particles.

Hilgers does not expressly state that the gaseous reaction products are cooled in a tubular heat exchanger. However, it would have been obvious to use a tubular heat exchanger to cool the gas phase products of the reaction required by Hiligers, before recycling them to the reactor via a pipe (Column 9, Example 3).

Hilgers does not expressly state the recycled gaseous reaction products are injected at a pressure of less than 75 psig above the reactor pressure. However, it would have been obvious to return the gaseous product to the reactor near to the pressure of the reactor, to avoid the excess cost of pressuring the recycle stream (Column 9, Example 3).

With respect to claim 8, Hilgers does not expressly state that the quench fluid has been cooled sufficiently to transform to a liquid phase prior to injecting into the reactor. However, it would have been obvious to cool the reaction products of Hilgers sufficiently to transform to a liquid phase prior to injecting into the reactor in order to store the gaseous product more efficiently in smaller containers before being recycled to the reactor.

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With respect to claim 11 and 19, Hilgers discloses that the quench fluid is injected into the reactor at a point downstream of the point in the reactor where oxygen and titanium tetrachloride are first reacted.

Hilgers does not expressly state the length between the point where oxygen and titanium tetrachloride are first reacted and the point in which the quench fuel is injected. However, where the only difference between the prior art and the claims is a recitation of relative dimensions of the claimed device and the device having the claimed relative dimensions would not perform differently than the prior art device, the claimed device is not patentably distinct from the prior art device. *In Gardner v. TEC Systems, Inc.*, 725 F.2d 1338, 220 USPQ 777 (Fed. Cir. 1984), cert. denied, 469 U.S. 830, 225 USPQ 232 (1984).

Claims 1-19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hilgers '168 in view of Yuill '893.

Hilgers does not expressly state that the quenched titanium dioxide particles and gaseous reaction products are made to follow a spiral path as they flow through the tubular heat exchanger.

Yuill teaches the quenched titanium dioxide particles and gaseous reaction products are made to follow a spiral path as they flow through the tubular heat exchanger (Column 1). Hilgers and Yuill are analogous art because they are from the same field of endeavor, namely titanium oxide processes.

At the time of invention it would have been obvious to a person of ordinary skill in the art to form the process of Hilgers to include the quenched titanium dioxide particles Art Unit: 1754

and gaseous reaction products are made to follow a spiral path as they flow through the tubular heat exchanger in view of the teaching of Yuill. The suggestion or motivation for doing so would have been to provide a means of cooling the reaction products as required by Hilgers, and to increase the removal of the deposits from the surfaces of the heat exchanger thereby increasing the heat transfer efficiency of the heat exchanger (Column 1).

Conclusion

The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Dunham '152 teaches vapor phase production of titanium dioxide pigments. Rado '090 teaches a method for processing gaseous effluent streams recovered from the vapor phase oxidation of metal halides. Boulos '153 discloses a process of producing titanium dioxide by a step of quenching with an inert gas downstream from the initial reaction point. Belknap '148 discloses a process of making titanium dioxide that includes a step of quenching the products downstream from the initial point of reaction.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to James A. Fiorito whose telephone number is (571)272-7426. The examiner can normally be reached on 9am - 6pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Stanley Silverman can be reached on (571) 272-1358. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

James Fiorito

Patent Examiner

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Primary Patent Examiner

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